

1,4-Bis(2-phenylethoxy)benzene

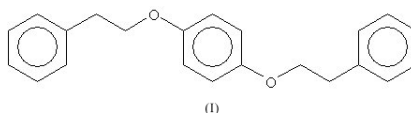
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Key indicators

Single-crystal X-ray study
 $T = 293\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$
 R factor = 0.045
 wR factor = 0.131
Data-to-parameter ratio = 20.0For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.The title compound, $\text{C}_{22}\text{H}_{22}\text{O}_2$, crystallizes in the space group $P2_1/c$, with one half molecule in the asymmetric unit and $Z = 2$.Received 24 April 2003
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Comment

The title compound, (I), is a new compound, synthesized as a precursor of substituted phenylene–vinylene oligomers, which are novel compounds for use as semiconductors in an electronic nose (Vanneste *et al.*, 1998; De Wit *et al.*, 1998). The molecule sits on a centre of symmetry. The structure of (I) is, apart from the outer phenyl rings, planar, with a maximum deviation in the torsion angles of $-176.74(13)^\circ$ for $\text{C1}-\text{O1}-\text{C4}-\text{C5}$. The dihedral angle between the least-squares planes through the central and each of the phenyl rings is $79.38(9)^\circ$. The particular arrangement of the rings allows an intermolecular interaction between the π -system of the outer rings and one of the H atoms of the central ring [$\text{Cg1} \cdots \text{H2}^1$ 2.71 Å; Cg1 is the centroid of ring $\text{C6}-\text{C11}$; symmetry code: (i) $1 - x, \frac{1}{2} - y, \frac{1}{2} + z$]. Other than that, the structure contains no unusual features.



Experimental

The synthesis of (I) was carried out under a nitrogen atmosphere. Sodium (13.8 g, 0.60 mol) in ethanol (100 ml) was added dropwise to a stirred solution of hydroquinone (33.0 g, 0.30 mol) in ethanol (100 ml). 2-Bromoethylbenzene (84.6 ml, 0.62 mol) was added dropwise to the reaction mixture, which was stirred overnight at room temperature. The temperature was subsequently raised to 318 K for a duration of 7 h, and finally to 353 K overnight. The solvent was evaporated on a rotary evaporator and the resulting oil dissolved in 50 ml of diethyl ether. The ether fraction was washed three times with 50 ml of a 10% NaOH solution and three times with 30 ml of water. After the solution was dried over MgSO_4 , the solvent was distilled off. After cooling, the resulting oil solidified. Recrystallization from ethanol yielded 11.6 g of 1,4-bis(2-phenylethoxy)benzene, which

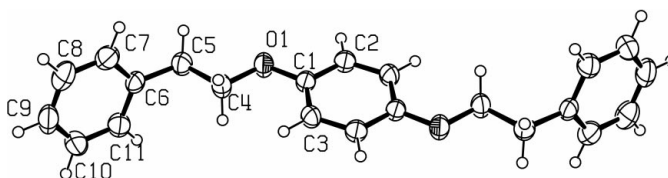


Figure 1

The molecular structure of (I), showing 50% probability displacement ellipsoids; the atom-numbering scheme of the asymmetric unit is shown.

crystallized as white needles. The final yield was 13%. M.p. 335 K; ^1H NMR (400 MHz, CDCl_3 , TMS): δ 7.32–7.18 (10H, *m*, H7, H11, H9, H8, H10), 6.80 (4H, *s*, H2, H3), 4.10 (2H, *t*, $J = 7.17$ Hz, H41, H42), 3.05 (2H, *t*, $J = 7.10$ Hz, H51, H52); ^{13}C NMR (100 MHz, CDCl_3 , TMS): δ 153.10 (C1), 138.38 (C6), 128.94 (C7, C11), 128.43 (C8, C10), 126.39 (C9), 115.63 (C2), 69.42 (C4), 35.93 (C5).

Crystal data

$\text{C}_{22}\text{H}_{22}\text{O}_2$	$D_x = 1.202 \text{ Mg m}^{-3}$
$M_r = 318.40$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 25 reflections
$a = 8.9286$ (17) Å	$\theta = 5\text{--}16^\circ$
$b = 11.3470$ (11) Å	$\mu = 0.08 \text{ mm}^{-1}$
$c = 8.925$ (2) Å	$T = 293$ (2) K
$\beta = 103.311$ (16) $^\circ$	Block, colourless
$V = 879.9$ (3) Å 3	$0.3 \times 0.3 \times 0.2 \text{ mm}$
$Z = 2$	

Data collection

Enraf–Nonius MACH3 diffractometer	$\theta_{\text{max}} = 32.0^\circ$
Non-profiled $\omega/2\theta$ scans	$h = -13 \rightarrow 12$
Absorption correction: none	$k = -16 \rightarrow 16$
6278 measured reflections	$l = 0 \rightarrow 13$
3053 independent reflections	3 standard reflections
1318 reflections with $I > 2\sigma(I)$	frequency: 60 min
$R_{\text{int}} = 0.083$	intensity decay: 1%

Refinement

Refinement on F^2	All H-atom parameters refined
$R[F^2 > 2\sigma(F^2)] = 0.045$	$w = 1/[\sigma^2(F_o^2) + (0.0591P)^2]$
$wR(F^2) = 0.131$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 0.99$	$(\Delta/\sigma)_{\text{max}} = 0.003$
3053 reflections	$\Delta\rho_{\text{max}} = 0.14 \text{ e \AA}^{-3}$
153 parameters	$\Delta\rho_{\text{min}} = -0.20 \text{ e \AA}^{-3}$

The low proportion of ‘observed’ reflections is due to the high temperature movement of the atoms at room temperature, causing the average reflection between 28 and 32° to have an $I/\sigma(I)$ of only 1.53.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms, 1996); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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